

Acta Crystallographica Section C

**Crystal Structure
Communications**

ISSN 0108-2701

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Electronic paper

This paper is published electronically. It meets the data-validation criteria for publication in Acta Crystallographica Section C. The submission has been checked by a Section C Co-editor though the text in the 'Comments' section is the responsibility of the authors.

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5-Fluoro-1-octanoyluracil

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Received 6 September 2000

Accepted 20 September 2000

Data validation number: IUC0000266

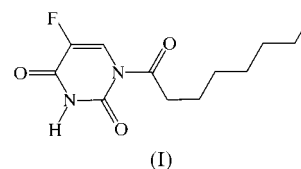
The crystal structure of 5-fluoro-1-octanoyluracil [5-fluoro-1-octanoylpyrimidine-2,4(1*H*,3*H*)-dione, C₁₂H₁₇FN₂O₃], a lipophilic prodrug of 5-fluorouracil, is described. The 5-fluoropyrimidine-2,4(1*H*,3*H*)-dione moiety is similar to the known structure of 1-acetyl-5-fluorouracil. The 1-octanoyl group and the 5-fluorouracil moiety are essentially coplanar, with the octanoyl carbonyl group oriented towards the the ring C—H group and away from the nearer ring carbonyl group. The torsion angle C—N—C—O (from the ring CH group to the octanoyl carbonyl group) of 9.2 (2)° is similar to the corresponding torsion angles reported for 1-acetyl-5-fluorouracil (17.3 and 1.6°) and 1,3-diacetyl-5-fluorouracil (8.8°).

Comment

The antimetabolite 5-fluorouracil is used for the treatment of solid tumors such as gastrointestinal adenocarcinoma, breast cancer and squamous cell carcinoma of the head and neck (Iyer & Ratain, 1999). 5-Fluorouracil cannot be administered orally because of its unpredictable absorption, non-linear pharmacokinetics and a high interpatient variance. Numerous 5-fluorouracil analogues have been synthesized to improve the delivery of 5-fluorouracil (Iyer & Ratain, 1999; Lamont & Schilsky, 1999; Ozaki, 1996). After delivery to the target tissue, these analogues are subject to chemical or enzymatic hydrolysis *in vivo* and release 5-fluorouracil (Bundgaard *et al.*, 1983; Møllgaard *et al.*, 1982). There is currently growing interest in lipophilic 1- and 3-acyl derivatives for transdermal drug delivery. Despite this potential application of acyl-5-fluorouracil derivatives, only the crystal structures of two acyl derivatives, namely 1-acetyl- and 1,3-diacetyl-5-fluorouracil, have been described (Beall *et al.*, 1993).

The structures of both 1-acetyl- and 1-octanoyl-5-fluorouracil, *i.e.* the 5-fluoropyrimidine-2,4(1*H*,3*H*)-dione system, are very similar. The 1-octanoyl group and the 5-fluorouracil moiety of the title compound, (I), are essentially coplanar, with the C7=O7 carbonyl group oriented towards the C6—H group and away from the C2=O2 group. The torsion angle C6—N1—C7—O7 is 9.2 (2)° and is similar to the torsion

angles reported for 1-acetyl-5-fluorouracil (17.3 and 1.6°) and 1,3-diacetyl-5-fluorouracil (8.8°) (Beall *et al.*, 1993). Most likely, the slight differences are due to packing effects in the crystal. Thus, the carbonyl of the 1-acyl group can be conjugated with the pyrimidine-2,4(1*H*,3*H*)-dione ring system. As a result of the orientation of the acyl group, the partially positive carbonyl C7 atom is easily accessible to nucleophiles such as hydroxide, and the hydrolysis of 1-acyl-5-fluorouracil derivatives is fast. For example, the half-life of 1-acetyl-5-fluorouracil is about 4.8 min (Beall *et al.*, 1993).



Experimental

5-Fluoro-1-octanoyluracil was synthesized by acylation of 5-fluorouracil with octanoyl chloride (Roberts & Sloan, 1999; Taylor & Sloan, 1998). White crystals were obtained upon crystallization from diethyl ether at 253 K (m.p. 336–338 K). ¹H NMR (CDCl₃, 400 MHz): δ 0.88 (*t*, —CH₃, *J* = 6.8 Hz, 3H), 1.20–1.42 (*m*, 8H), 1.72 (*q*, —CH₂CH₂CON, *J* = 7.4 Hz, 2H), 3.12 (*t*, —CH₂CON, *J* = 7.4 Hz, 2H), (*d*, —CH=CF—, *J* = 6.8 Hz, 1H); ¹³C NMR (CDCl₃, 75 MHz): δ 14.04 (—CH₃), 22.56, 24.37, 28.84, 28.94, 31.59, 39.04, 121.75 (=CH—, *J*_{CF} = 27 Hz), 141.26 (=CF—, *J*_{CF} = 182 Hz), 147.75 (N—CO—NH), 156.62 (=CF—CO—NH, *J*_{CF} = 21 Hz), 171.98 (acyl CO); ¹⁹F NMR (CDCl₃): δ —161.10 (*d*, 6.0 Hz); IR (cm^{−1}): 1738 and 1704 [ν(C=O)]; MS *m/z* (relative intensity, %): 256 (1, *M*⁺), 127 (100, C₈H₁₅O⁺), 57 (84), 43 (31).

Crystal data

C ₁₂ H ₁₇ FN ₂ O ₃	<i>D</i> _x = 1.347 Mg m ^{−3}
<i>M</i> _r = 256.28	Mo <i>K</i> α radiation
Triclinic, <i>P</i> 1̄	Cell parameters from 4236 reflections
<i>a</i> = 5.4500 (11) Å	<i>θ</i> = 1.00–25.35°
<i>b</i> = 9.7410 (19) Å	<i>μ</i> = 0.107 mm ^{−1}
<i>c</i> = 12.307 (3) Å	<i>T</i> = 173 (1) K
<i>α</i> = 80.27 (3)°	Irregular plate-like fragment, colourless
<i>β</i> = 85.97 (3)°	0.32 × 0.20 × 0.04 mm
<i>γ</i> = 79.13 (3)°	
<i>V</i> = 631.9 (2) Å ³	
<i>Z</i> = 2	

Data collection

Nonius KappaCCD diffractometer	<i>θ</i> _{max} = 25.23°
<i>ω</i> scans at fixed <i>χ</i> = 55°	<i>h</i> = −6 → 6
4358 measured reflections	<i>k</i> = −11 → 11
2277 independent reflections	<i>l</i> = −14 → 14
1562 reflections with <i>I</i> > 2σ(<i>I</i>)	Intensity decay: <1%
<i>R</i> _{int} = 0.034	

Refinement

Refinement on <i>F</i> ²	<i>w</i> = 1/[σ ² (<i>F</i> _o ²) + (0.0424 <i>P</i>) ²]
<i>R</i> [<i>F</i> ² > 2σ(<i>F</i> ²)] = 0.041	where <i>P</i> = (<i>F</i> _o ² + 2 <i>F</i> _c ²)/3
<i>wR</i> (<i>F</i> ²) = 0.100	(Δ/σ) _{max} < 0.001
<i>S</i> = 1.046	Δρ _{max} = 0.21 e Å ^{−3}
2277 reflections	Δρ _{min} = −0.20 e Å ^{−3}
165 parameters	Extinction correction: <i>SHELXL97</i>
H-atom parameters constrained	Extinction coefficient: 0.048 (5)

Table 1

Selected geometric parameters (Å, °).

N1—C6	1.3950 (19)	C5—F5	1.3490 (18)
N1—C2	1.410 (2)	O7—C7	1.2084 (18)
N1—C7	1.449 (2)	C7—C8	1.500 (2)
C2—O2	1.2074 (18)	C8—C9	1.527 (2)
C2—N3	1.3766 (19)	C9—C10	1.517 (2)
N3—C4	1.3714 (19)	C10—C11	1.523 (2)
C4—O4	1.2282 (18)	C11—C12	1.520 (2)
C4—C5	1.440 (2)	C12—C13	1.514 (2)
C5—C6	1.316 (2)	C13—C14	1.517 (2)
C6—N1—C2	120.36 (14)	F5—C5—C4	116.15 (14)
C6—N1—C7	115.90 (13)	C5—C6—N1	121.15 (15)
C2—N1—C7	123.71 (13)	O7—C7—N1	116.73 (14)
O2—C2—N3	121.57 (14)	O7—C7—C8	123.48 (15)
O2—C2—N1	124.02 (15)	N1—C7—C8	119.78 (14)
N3—C2—N1	114.41 (14)	C7—C8—C9	111.77 (14)
C4—N3—C2	128.42 (14)	C10—C9—C8	112.19 (13)
O4—C4—N3	122.42 (15)	C9—C10—C11	114.19 (14)
O4—C4—C5	125.06 (16)	C12—C11—C10	113.52 (14)
N3—C4—C5	112.53 (15)	C13—C12—C11	114.28 (14)
C6—C5—F5	120.89 (14)	C12—C13—C14	113.15 (14)
C6—C5—C4	122.95 (16)		

Data collection: *COLLECT* (Nonius, 1998); cell refinement: *SCALEPACK* (Otwinowski & Minor, 1997); data reduction: *DENZO-SMN* (Otwinowski & Minor, 1997); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); software used to

prepare material for publication: *SHELXL97* (Sheldrick, 1997) and local programs.

HJL would like to thank Dr L. W. Robertson for providing support and laboratory facilities and Dr J. Goodman from the University of Kentucky Life Sciences Mass Spectrometry Facility for performing the mass spectral analysis. This work was supported in part by the University of Kentucky Medical Center Research Fund.

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